Safety Data Sheet

according to Federal Register / Vol. 77, No. 58 / Monday, March 26, 2012 / Rules and RegulationsDate of issue: 04/18/2014Revision date: 04/09/2021Supersedes: 08/01/2019

SECTION 1: Identification					
1.1. Identification					
Product form	: Substance				
Trade name	: UtiliMag® 40				
Chemical name	: Magnesium oxide				
CAS-No.	: 1309-48-4				
Formula	: MgO			"	
Other means of identification		ia, calcined magnesia, calc periclase, sea-water magne			
1.2. Recommended use and restriction	ns on use				
Use of the substance/mixture	: For use in utility application treatment.	ons such as air scrubbing,	acid neutr	tralization and backend boiler	
1.3. Supplier					
Martin Marietta Magnesia Specialties					
1800 Eastlake Road					
Manistee, Michigan 49660 - USA					
T +001 410 780 5500					
1.4. Emergency telephone number Emergency number	: CHEMTREC, U.S.: 1-80()-424-9300 INTERNATIO	NAL: +1-7	703-527-3887 Available 24/7	
SECTION 2: Hazard(s) identification					
2.1. Classification of the substance or	mixture				
GHS US classification					
Not classified					
					_
2.2. GHS Label elements, including pre	ecautionary statements				
GHS US labeling					
No labeling applicable					
2.3. Other hazards which do not result	t in classification				
Other hazards not contributing to the classification	: No additional hazards h	ave been identified.			None
2.4. Unknown acute toxicity (GHS US)					
Not applicable					
SECTION 3: Composition/Informati	ion on ingredients				
3.1. Substances					
Substance type	: Mono-constituent				
Name	: UtiliMag Magnesium Oxid	1e			
CAS-No.	: 1309-48-4	5			
Name		Product identifier	%	GHS US classification	
Name Magnesium oxide		(CAS-No.) 1309-48-4	% 98	Not classified	-
Oxides of silicon, iron, aluminum, and calcium		(CAS-No.) 1309-46-4 (CAS-No.) mixture	90 2	Not classified	—
			-		
3.2. Mixtures					
Not applicable					
SECTION 4: First-aid measures					
4.1. Description of first aid measures					
First-aid measures general	 Never give anything by m advice (show the label wh 		erson. If ye	vou feel unwell, seek medical	
First-aid measures after inhalation	,	g is difficult, remove victim	to fresh ai	air and keep at rest in a position	
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First-aid measures after skin contact	Not expected to be an irritant. Remove affected clothing and wash all exposed skin area with mild soap and water, followed by warm water rinse.	
First-aid measures after eye contact	: Rinse immediately with plenty of water. Obtain medical attention if pain, blinking or redness persists.	
First-aid measures after ingestion	: Rinse mouth. Do NOT induce vomiting.	
4.2. Most important symptoms and effect	ts (acute and delayed)	
Symptoms/effects	Not expected to present a significant hazard under anticipated conditions of normal use. Do not breathe dust.	
Symptoms/effects after inhalation	Inhalation may cause: irritation, cough, shortness of breath.	
Symptoms/effects after skin contact	: None under normal conditions.	
Symptoms/effects after eye contact	: May cause eye irritation.	
Symptoms/effects after ingestion	: Ingestion generally causes purging of the bowels. Swallowing large amounts may cause bowel obstruction.	

4.3. Immediate medical attention and special treatment, if necessary

No special procedures required.

SECTION 5: Fire-fighting measures				
5.1. Suitable (and unsuitable) extinguishing media				
Suitable extinguishing media	: Not combustible. If there is a fire close by, use suitable extinguishing agents. Water fog. Carbon dioxide. Dry powder. Foam.			
Unsuitable extinguishing media	: None known.			
5.2. Specific hazards arising from the ch	emical			
Fire hazard	If heated to decomposition (>1700 °C), magnesium oxide fumes may be generated.			
Explosion hazard	Product is not explosive.			
Reactivity	: Reacts with : Incompatible materials.			
5.3. Special protective equipment and precautions for fire-fighters				
Firefighting instructions	: Use water spray or fog for cooling exposed containers. Exercise caution when fighting any chemical fire. Do not allow run-off from fire fighting to enter drains or water courses.			
Protection during firefighting	: Do not enter fire area without proper protective equipment, including respiratory protection.			
Other information	: No additional risk management measures required.			

SECTION 6: Accidental release measures					
6.1. Personal precautions, protecti	6.1. Personal precautions, protective equipment and emergency procedures				
General measures	: Avoid creating or spreading dust. Dust deposited may be vacuum cleaned.				
6.1.1. For non-emergency personnel					
Protective equipment	: Where excessive dust may result, use approved respiratory protection equipment.				
Emergency procedures	: Evacuate unnecessary personnel.				
6.1.2. For emergency responders					
Protective equipment	: Where excessive dust may result, use approved respiratory protection equipment.				
Emergency procedures	: Sweep spilled substance into containers; if appropriate, moisten first to prevent dusting.				
6.2. Environmental precautions					
Prevent entry to sewers and public waters.					
6.3. Methods and material for conta	ainment and cleaning up				
For containment	: Contain and collect as any solid.				
Methods for cleaning up	: Sweep up spilled material without making dust.				
6.4. Reference to other sections					
See Heading 8. Exposure controls and per	sonal protection.				
SECTION 7: Handling and storage	ge				
7.1 Propoutions for sofe handling					

7.1. Precautions for sale handling	
Precautions for safe handling	: Provide good ventilation in process area to prevent formation of dust.
Hygiene measures	: Handle in accordance with good industrial hygiene and safety procedures.

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7.2.	. Conditions for safe storage, including any incompatibilities	
Storage conditions : Keep container closed when not in use.		
Incomp	batible materials	 ACID (Strong) - vigorous reaction, heat generated; Chlorine Trifluoride reacts violently, producing flame; Phosphorous Pentachloride - incandesces brilliantly. NOTE: Exposure to water may cause this product to slowly hydrate, during which heat may be generated (exothermic reaction).
Prohibi	tions on mixed storage	: Keep away from incompatible materials.

SECTION 8: Exposure controls/personal protection 8.1. Control parameters

on onto parameters				
UtiliMag® 40 (1309-48-4)				
ACGIH	Local name	Magnesium oxide		
ACGIH	ACGIH TWA (mg/m ³)	10 mg/m ³		
Magnesium oxide (1	1309-48-4)			
ACGIH	Local name	Magnesium oxide		
ACGIH ACGIH TWA (mg/m ³)		10 mg/m ³		
ACGIH Remark (ACGIH)		(inhalable fraction)		
ACGIH Regulatory reference ACGIH 2019				
OSHA OSHA PEL (TWA) (mg/m ³) 15 mg/m ³				
Oxides of silicon, iron, aluminum, and calcium (mixture)				
Not applicable				

8.2. Appropriate engineering controls

Appropriate engineering controls

: Provide local exhaust or general room ventilation to minimize exposure to dust. Avoid dispersal of dust in the air (i.e, clearing dust surfaces with compressed air).

8.3. Individual protection measures/Personal protective equipment

Eye protection:

Safety glasses with side guards should be worn to prevent injury from airborne particles and/or other eye contact with this product. Where excessive dust may result, wear goggles

Respiratory protection:

In case of inadequate ventilation wear respiratory protection. Use an N95 respirator.

SECTION 9: Physical and chemical properties			
9.1. Information on basic physical and	chemical properties		
Physical state	: Solid		
Appearance	: Powder.		
Color	: white		
Odor	: odorless		
Odor threshold	: No data available		
рН	: No data available		
pH solution	: 10.3 saturated aqueous solution		
Melting point	: 2827 (2797 - 2857) °C		
Freezing point	: No data available		
Boiling point	: 3600 °C		
Flash point	: Product does not sustain combustion		
Relative evaporation rate (butyl acetate=1)	: No data available		
Flammability (solid, gas)	: Non flammable.		
Vapor pressure	: No data available		
Vapor pressure at 50 °C	: 0 hPa		
Relative vapor density at 20 °C	: 0		
Flammability (solid, gas) Vapor pressure Vapor pressure at 50 °C	 Non flammable. No data available 0 hPa 		

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Relative density	: No data available	
Specific gravity / density	: 3.58 g/cm ³ (theoretical density of MgO)	
Molecular mass	: 40.3 g/mol	
Solubility	: In water, material is partially soluble.	
Log Pow	: No data available	
Auto-ignition temperature	: No data available	
Decomposition temperature	: >1700 °C	
Viscosity, kinematic	: No data available	
Viscosity, dynamic	: No data available	
Explosion limits	: No data available	
Explosive properties	: Product is not explosive.	
Oxidizing properties	: No data available	
9.2. Other information		

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SECT	ION 10: Stability and reactivity
10.1.	Reactivity
Reacts	with : Incompatible materials.
10.2.	Chemical stability
Stable a	at ambient temperature and under normal conditions of use.
10.3.	Possibility of hazardous reactions
Hazardo	ous polymerization will not occur.
10.4.	Conditions to avoid
Keep/S	tore away from incompatible materials.
10.5.	Incompatible materials
ACID (S	Strong) - vigorous reaction, heat generated; Chlorine Trifluoride reacts violently, producing flame; Phosphorous Pentachloride - incandesces

ACID (Strong) - vigorous reaction, heat generated; Chlorine Trifluoride reacts violently, producing flame; Phosphorous Pentachloride - incandeso brilliantly. NOTE: Exposure to water may cause this product to slowly hydrate, during which heat may be generated (exothermic reaction).

10.6. Hazardous decomposition products

If magnesium oxide is heated to the point of volatilization (i.e, >1700 C), magnesium oxide fumes may be generated.

SECTION 11: Toxicological information

11.1.	Information on toxicological effects		
Acute to	xicity (oral)	:	Not classified (Based on available data, the classification criteria are not met)
Acute to	xicity (dermal)	:	Not classified
Acute to	vicity (inhalation)	:	Not classified

Magnesium oxide (1309-48-4)		
LD50 oral rat	3870 - 3990 mg/kg	
ATE US (oral)	3870 mg/kg body weight	
Skin corrosion/irritation	: Not classified (Based on available data, the classification criteria are not met)	
Serious eye damage/irritation	: Not classified (Based on available data, the classification criteria are not met)	
Respiratory or skin sensitization	: Not classified (Based on available data, the classification criteria are not met)	
Germ cell mutagenicity	: Not classified (Based on available data, the classification criteria are not met)	
Carcinogenicity	: Not classified (Based on available data, the classification criteria are not met)	
Reproductive toxicity	: Not classified (Based on available data, the classification criteria are not met)	
Specific target organ toxicity - single exposure	: Not classified (Based on available data, the classification criteria are not met)	
Specific target organ toxicity – repeated exposure	: Not classified (Based on available data, the classification criteria are not met)	
Aspiration hazard	: Not classified (Based on available data, the classification criteria are not met)	
Viscosity, kinematic	: No data available	
Likely routes of exposure	: dermal. Inhalation.	

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Symptoms/effects	: Not expected to present a significant hazard under anticipated conditions of normal use. Do not breathe dust.
Symptoms/effects after inhalation	: Inhalation may cause: irritation, cough, shortness of breath.
Symptoms/effects after skin contact	: None under normal conditions.
Symptoms/effects after eye contact	: May cause eye irritation.
Symptoms/effects after ingestion	 Ingestion generally causes purging of the bowels. Swallowing large amounts may cause bowel obstruction.

SECTION 12: Ecological information

12.1. IOXICITY			
Magnesium oxide (1309-48-4)			
LC50 fish 1	1355 mg/l		
EC50 crustacea	190 mg/l		
12.2. Persistence and degradability UtiliMag® 40 (1309-48-4)			
Persistence and degradability	Not established.		
12.3. Bioaccumulative potential			
UtiliMag® 40 (1309-48-4)			

Othimag® 40 (1509-46-4)	
Bioaccumulative potential	Not established.

12.4. Mobility in soil

No additional information available

12.5. Other adverse effects

Other information

: Avoid release to the environment.

SECTION 13: Disposal considerations			
13.1. Disposal methods			
Waste disposal recommendations Ecology - waste materials	Dispose in a safe manner in accordance with local/national regulations.Avoid release to the environment.		

SECTION 14: Transport information

Department of Transportation (DOT)

In accordance with DOT

Not regulated.

Transport by sea

Not regulated.

Air transport

Not regulated.

SECTION 15: Regulatory information

15.1. US Federal regulations

Magnesium oxide (1309-48-4)			
Listed on the United States TSCA (Toxic Substances Control Act) inventory			
SARA Section 311/312 Hazard Classes	Immediate (acute) health hazard	No	
	Delayed (chronic) health hazard	No	
	Fire hazard	No	
	Sudden release of pressure hazard	No	

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Magnesium oxide (1309-48-4)		
	Reactive hazard	No
SARA Section 313 - Emission Reporting	Magnesium oxide is not hazardous and is not subject to Form R reporting requirements.	

15.2. International regulations

Magnesium oxi		
Jurisdiction	List	Comment
Asia Pacific	Asia - PAC	
Australia	Australian Inventory of Chemical Substances (AICS)	
	National Pollutant Inventory	magnesium oxide fume
China	Priority Existing Chemicals Inventory of Existing Chemical Substances (IECSC)	
Japan	Existing and New Chemical Substances (ECSC)	# 1-465; inorganic compounds
Korea	KECI (Chemical Inventory of Korea)	KE-22728
New Zealand	Inventory of Chemicals (NZIoC)	HSNO approval
Phillippines	Inventory of Chemicals and Chemical Substances (PICCS)	
Europe	EEC International Cosmetics Ingredients Inventory (INCI)	absorbant/ buffering/ opacifying / additives
	EU REACH pre-registered	
	EU Inventory of Existing Commercial Chemical Substances (EINECS)	215-171-9
	German Water Hazard Class Substance List	5208
		Classification: VwVwS
	Switzerland Giftliste 1 (List of Toxic Substances)	G-2368
Canada	Canadian Domesticated Substances List (DSL)	
	WHMIS Ingredient List	
United States	ACGIH Thrshold Limit Values (TLV)	
	EPA Pesticide Inert Ingredients	
	FDA Priority-based Assessment of Food Additives (PAFA)	
	FDA Regulations	Use as colorant.
	High Production Volume Chemicals (HPV)	
	National Toxicology Program Technical Reports List	
	NIOSH Hazard, Toxicology, and Use Information	
	NIOSH Health Hazards	
	NIOSH Recommended Exposure Limits	10 mg/m ³
	OSHA Permissible Exposure Limits	8 hour TWA: total particulates 15 mg/ m ³
	Toxic Substances Control Act (TSCA) Inventory	
	Toxic Inventory Update Rule	
	TSCA Section 8A-Preliminary Assessment Information Rule (PAIR)	
Other	Health Hazards	RTECS: OM3850000
	High Production Volume Chemicals: ICCA	
	High Production Volume Chemicals: OECD	

15.3. US State regulations

WARNIN :

This product can expose you to Lead and Nickel compounds, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

Component	State or local regulations
Magnesium oxide(1309-48-4)	U.S Idaho - Non-Carcinogenic Toxic Air Pollutants - Acceptable Ambient Concentrations; U.S New Jersey - Right to Know Hazardous Substance List; U.S Pennsylvania - RTK (Right to Know) List

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SECTION 16: Other information

according to Federal Register /	Vol. 77, No. 58 / Monday, March 26, 2012 / Rules and Regulations
Revision date	04/09/2021
Data sources	ACGIH 2019.
	Chemical Inspection & Regulation Service; accessed at: <u>http://www.cirs-</u> reach.com/Inventory/Global_Chemical_Inventories.html.
	Ind. Exposure & Control Techn. for OSHA Regulated Substances - MgO (fume), March, 1989, pp. 1181-1184.
	Krister Forsberg and S.Z. Mansdorf, "Quick Selection Guide to Chemical Protective Clothing", Fifth Edition.
	NIOSH Occupational Health Guide for chemical Substances - Vol. II, September, 1978.
	REGULATION (EC) No 1272/2008 OF THE EUROPEAN PARLIAMENT AND OF THE COUNCIL of 16 December 2008 on classification, labelling and packaging of substances and mixtures, amending and repealing Directives 67/548/EEC and 1999/45/EC, and amending Regulation (EC) No 1907/2006.
	RTECS, June 1998.
	Sax - 8th Ed. TSCA Chemical Substance Inventory. Accessed at
	http://www.epa.gov/oppt/existingchemicals/pubs/tscainventory/howto.html
	US National Library of Medicine National Institutes of Health Haz-Map. Accessed at http://hazmap.nlm.nih.gov
Other information	None.
Abbreviations and acronym	s: ACGIH (American Conference of Government Industrial Hygienists)
	ATE: Acute Toxicity Estimate
	CAS (Chemical Abstracts Service) number
	EC50: Environmental Concentration associated with a response by 50% of the test population.
	GHS: Globally Harmonized System (of Classification and Labeling of Chemicals
	LD50: Lethal Dose for 50% of the test population
	OSHA: Occupational Safety & Health Administration
	TSCA: Toxic Substances Control Act
	TWA: Time Weighted Average
NFPA health hazard	: 0 - Materials that, under emergency conditions, would offer no hazard beyond that of ordinary combustible materials.
NFPA fire hazard	: 0 - Materials that will not burn under typical fire conditions, including intrinsically noncombustible materials such as concrete, stone, and sand.
NFPA reactivity	: 0 - Material that in themselves are normally stable, even under fire conditions.

Indication of changes:

Section	Changed item	Change	Comments
15	California Proposition 65 Disclosure	Added	
SDS Prepared by:	The Redstone Group 6077 Frantz Rd. Suite 206 Dublin, Ohio, USA 43017 614.923.7472 www.redstonegrp.com		

This information is based on our current knowledge and is intended to describe the product for the purposes of health, safety and environmental requirements only. It should not therefore be construed as guaranteeing any specific property of the product.