Safety Data Sheet

according to Federal Register / Vol. 77, No. 58 / Monday, March 26, 2012 / Rules and Regulations

Date of issue: 04/18/2014 Revision date: 03/02/2021 Supersedes: 08/01/2019 Version: 1.3

SECTION 1: Identification

1.1. Identification

Product form : Mixture

Trade name : Versamag® 30

CAS-No. : 01309-42-8 / 01309-48-4

Formula : Mg(OH)2 / MgO

1.2. Recommended use and restrictions on use

Use of the substance/mixture : For use in specialty or industrial applications.

1.3. Supplier

Martin Marietta Magnesia Specialties

1800 Eastlake Road

Manistee, Michigan 49660 - USA

T +001 410 780 5500

1.4. Emergency telephone number

Emergency number : CHEMTREC, U.S.: 1-800-424-9300 INTERNATIONAL: +1-703-527-3887 Available 24/7

SECTION 2: Hazard(s) identification

2.1. Classification of the substance or mixture

GHS US classification

Not classified

2.2. GHS Label elements, including precautionary statements

GHS US labeling

No labeling applicable

2.3. Other hazards which do not result in classification

Other hazards not contributing to the

: No additional hazards have been identified.

classification

2.4. Unknown acute toxicity (GHS US)

Not applicable

SECTION 3: Composition/Information on ingredients

3.1. Substances

Not applicable

3.2. Mixtures

Name	Product identifier	%	GHS US classification
Magnesium hydroxide	(CAS-No.) 1309-42-8	> 50	Not classified
Magnesium oxide	(CAS-No.) 1309-48-4	< 50	Not classified
Oxides of silicon, iron, aluminum, and calcium	(CAS-No.) mixture	0.7	Not classified

SECTION 4: First-aid measures

4.1. Description of first aid measures

First-aid measures general : Never give anything by mouth to an unconscious person. If you feel unwell, seek medical

advice (show the label where possible).

First-aid measures after inhalation : If inhaled and if breathing is difficult, remove victim to fresh air and keep at rest in a position

comfortable for breathing.

First-aid measures after skin contact : Not expected to be an irritant. Remove affected clothing and wash all exposed skin area with

mild soap and water, followed by warm water rinse.

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First-aid measures after eye contact : Rinse immediately with plenty of water. Obtain medical attention if pain, blinking or redness

persists

First-aid measures after ingestion : Rinse mouth. Do NOT induce vomiting.

4.2. Most important symptoms and effects (acute and delayed)

Symptoms/effects : Not expected to present a significant hazard under anticipated conditions of normal use. Do not

breathe dust.

Symptoms/effects after inhalation : Inhalation may cause: irritation, cough, shortness of breath.

Symptoms/effects after skin contact : None under normal conditions. Symptoms/effects after eye contact : May cause eye irritation.

Symptoms/effects after ingestion : Ingestion generally causes purging of the bowels. Swallowing large amounts may cause bowel

obstruction.

4.3. Immediate medical attention and special treatment, if necessary

No special procedures required.

SECTION 5: Fire-fighting measures

5.1. Suitable (and unsuitable) extinguishing media

Suitable extinguishing media : Not combustible. If there is a fire close by, use suitable extinguishing agents. Water fog.

Carbon dioxide. Dry powder. Foam.

Unsuitable extinguishing media : None known.

5.2. Specific hazards arising from the chemical

Fire hazard : If magnesium hydroxide is heated to the point of decomposition (>350 °C), it forms magnesium

oxide and water. If magnesium oxide is heated to the point of volatilization (i.e, >1700 °C),

magnesium oxide fumes may be generated.

Explosion hazard : Product is not explosive.

Reactivity : Reacts with : Incompatible materials.

5.3. Special protective equipment and precautions for fire-fighters

Firefighting instructions : Use water spray or fog for cooling exposed containers. Exercise caution when fighting any

chemical fire. Do not allow run-off from fire fighting to enter drains or water courses.

Protection during firefighting : Do not enter fire area without proper protective equipment, including respiratory protection.

Other information : No additional risk management measures required.

SECTION 6: Accidental release measures

6.1. Personal precautions, protective equipment and emergency procedures

General measures : Avoid creating or spreading dust.

6.1.1. For non-emergency personnel

Protective equipment : Where excessive dust may result, use approved respiratory protection equipment.

Emergency procedures : Evacuate unnecessary personnel.

6.1.2. For emergency responders

Protective equipment : Where excessive dust may result, use approved respiratory protection equipment. Emergency procedures : Sweep spilled substance into containers; if appropriate, moisten first to prevent dusting.

6.2. Environmental precautions

Prevent entry to sewers and public waters.

6.3. Methods and material for containment and cleaning up

For containment : Contain and collect as any solid.

Methods for cleaning up : Sweep up spilled material without making dust.

6.4. Reference to other sections

See Heading 8. Exposure controls and personal protection.

SECTION 7: Handling and storage

7.1. Precautions for safe handling

Precautions for safe handling : Provide good ventilation in process area to prevent formation of dust.

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Hygiene measures

: Handle in accordance with good industrial hygiene and safety procedures. Wash hands and other exposed areas with mild soap and water before eating, drinking or smoking and when leaving work.

7.2. Conditions for safe storage, including any incompatibilities

Incompatible products

- : ACID (Strong) vigorous reaction, heat generated; MALEIC ANHYDRIDE Alkali and other alkaline earth compounds including magnesium compounds, will cause explosive decomposition of maleic anhydride; PHOSPHORUS – Phosphorus boiled with alkaline hydroxides yields mixed phosphines which may ignite spontaneously with air.
- : ACID (Strong) vigorous reaction, heat generated; Chlorine Trifluoride reacts violently, producing flame; Phosphorous Pentachloride incandesces brilliantly. NOTE: Exposure to water may cause this product to slowly hydrate, during which heat may be generated (exothermic reaction).

SECTION 8: Exposure controls/personal protection

8.1. Control parameters

Magnesium hydroxide (1309-42-8)			
ACGIH	ACGIH TWA (mg/m³)	10 mg/m³ as Particulates (insoluble or poorly soluble) not otherwise specified 3 mg/m³ (respirable fraction / fraction respirable)	
OSHA	OSHA PEL (TWA) (mg/m³)	10 mg/m³ (total dust) as inert or nuisance dust not otherwise regulated; 5 mg/m³ (respirable fraction) as inert or nuisance dust not otherwise regulated	
OSHA	OSHA PEL (TWA) (ppm)	15 mppcf	
OSHA	Regulatory reference (US-OSHA)	OSHA Annotated Table Z-3 Mineral Dusts	
Magnesium oxide (1309-48	Magnesium oxide (1309-48-4)		
ACGIH	Local name	Magnesium oxide	
ACGIH	ACGIH TWA (mg/m³)	10 mg/m³	
ACGIH	Remark (ACGIH)	(inhalable fraction)	
ACGIH	Regulatory reference	ACGIH 2019	
OSHA	OSHA PEL (TWA) (mg/m³)	15 mg/m³	
Oxides of silicon, iron, aluminum, and calcium (mixture)			
Not applicable			
Inorganic chloride salts (mixture)			
Not applicable			

Inorganic silicates and carbonates (mixture)

Not applicable

8.2. Appropriate engineering controls

Appropriate engineering controls

: Provide local exhaust or general room ventilation to minimize exposure to dust.

8.3. Individual protection measures/Personal protective equipment

Eye protection:

Safety glasses with side guards should be worn to prevent injury from airborne particles and/or other eye contact with this product. Where excessive dust may result, wear goggles

Respiratory protection:

In case of inadequate ventilation wear respiratory protection. Use an N95 respirator.

SECTION 9: Physical and chemical properties

9.1. Information on basic physical and chemical properties

Physical state : Solid
Appearance : Powder.
Color : white

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: No data available

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Odor : odorless

Odor threshold : No data available рΗ : No data available pH solution : ≥ 10 saturated solution Melting point : 350 °C decomposes Freezing point No data available Boiling point : No data available : No data available Flash point Relative evaporation rate (butyl acetate=1) : No data available Flammability (solid, gas) : Non flammable. Vapor pressure No data available Relative vapor density at 20 °C : No data available

Specific gravity / density : 2.9 g/cm³

Solubility : In water, material is partially soluble.

Log Pow : No data available

Auto-ignition temperature : Does not self-ignite

Decomposition temperature : > 350 °C

Viscosity, kinematic : No data available
Viscosity, dynamic : No data available
Explosion limits : No data available
Explosive properties : Product is not explosive.
Oxidizing properties : No oxidizing properties.

9.2. Other information

No additional information available

SECTION 10: Stability and reactivity

10.1. Reactivity

Relative density

Reacts with: Incompatible materials.

10.2. Chemical stability

Stable at ambient temperature and under normal conditions of use.

10.3. Possibility of hazardous reactions

Hazardous polymerization will not occur.

10.4. Conditions to avoid

Keep/Store away from incompatible materials.

10.5. Incompatible materials

ACID (Strong) - vigorous reaction, heat generated; MALEIC ANHYDRIDE – Alkali and other alkaline earth compounds including magnesium compounds, will cause explosive decomposition of maleic anhydride; PHOSPHORUS – Phosphorus boiled with alkaline hydroxides yields mixed phosphines which may ignite spontaneously with air.

10.6. Hazardous decomposition products

No additional information available

SECTION 11: Toxicological information

11.1. Information on toxicological effects

Acute toxicity (oral) : Not classified (Based on available data, the classification criteria are not met)

Acute toxicity (dermal) : Not classified Acute toxicity (inhalation) : Not classified

Magnesium hydroxide (1309-42-8)	
LD50 oral rat	> 2000 mg/kg OECD Guideline 423
Magnesium oxide (1309-48-4)	
LD50 oral rat	3870 - 3990 mg/kg
ATE US (oral)	3870 mg/kg body weight

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Skin corrosion/irritation : Not classified (Based on available data, the classification criteria are not met) Serious eye damage/irritation : Not classified (Based on available data, the classification criteria are not met) Respiratory or skin sensitization : Not classified (Based on available data, the classification criteria are not met) Germ cell mutagenicity : Not classified (Based on available data, the classification criteria are not met) Carcinogenicity : Not classified (Based on available data, the classification criteria are not met) Reproductive toxicity : Not classified (Based on available data, the classification criteria are not met) Specific target organ toxicity - single exposure : Not classified (Based on available data, the classification criteria are not met) Specific target organ toxicity - repeated : Not classified (Based on available data, the classification criteria are not met)

exposure

: Not classified (Based on available data, the classification criteria are not met) Aspiration hazard

Viscosity, kinematic : No data available : dermal. Inhalation. Likely routes of exposure

Symptoms/effects : Not expected to present a significant hazard under anticipated conditions of normal use. Do not

breathe dust.

Symptoms/effects after inhalation : Inhalation may cause: irritation, cough, shortness of breath.

Symptoms/effects after skin contact : None under normal conditions. Symptoms/effects after eye contact : May cause eye irritation.

Symptoms/effects after ingestion : Ingestion generally causes purging of the bowels. Swallowing large amounts may cause bowel

obstruction.

SECTION 12: Ecological information

Magnesium hydroxide (1309-42-8)	
LC50 fish 1	1293 mg/l Onchorinchus mykiss
EC50 crustacea	284.76 mg/l
LC50 fish 2	511.31 mg/l P. promelas
ErC50 (algae)	> 100 mg/l
Magnesium oxide (1309-48-4)	
LC50 fish 1	1355 mg/l
EC50 crustacea	190 mg/l

Persistence and degradability

Versamag® 30 (01309-42-8 / 01309-48-4)	
Persistence and degradability Not established.	
Magnesium hydroxide (1309-42-8)	
Persistence and degradability Not readily biodegradable.	
Biodegradation Does not degrade although it does dissolve.	

Bioaccumulative potential

No additional information available

Mobility in soil

No additional information available

Other adverse effects

Other information : Avoid release to the environment.

SECTION 13: Disposal considerations

Disposal methods

Waste disposal recommendations : Dispose in a safe manner in accordance with local/national regulations.

Ecology - waste materials : Avoid release to the environment.

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SECTION 14: Transport information

Department of Transportation (DOT)

In accordance with DOT

Not regulated.

Transport by sea

Not regulated.

Air transport

Not regulated.

SECTION 15: Regulatory information

Magnesium Hydroxide (1309-42-8) Magnesium oxide (1309-48-4)		
Listed on the United States TSCA (Toxic Substances Control Act) inventory		
SARA Section 311/312 Hazard Classes	Immediate (acute) health hazard	No
	Delayed (chronic) health hazard	No
	Fire hazard	No
	Sudden release of pressure hazard	No
	Reactive hazard	No
SARA Section 313 - Emission Reporting	Magnesium hydroxide and magnesium oxide reporting requirements.	are not hazardous and are not subject to Form R

15.2. International regulations

Magnesium oxid	de (1309-48-4)	
Jurisdiction	List	Comment
Asia Pacific	Asia - PAC	
Australia	Australian Inventory of Chemical Substances (AICS)	
	National Pollutant Inventory	magnesium oxide fume
	Priority Existing Chemicals	
China	Inventory of Existing Chemical Substances (IECSC)	
Japan	Existing and New Chemical Substances (ENCS)	# 1-465; inorganic compounds
Korea	KECI (Chemical Inventory of Korea)	KE-22728
New Zealand	Inventory of Chemicals (NZIoC)	HSNO approval
Phillippines	Inventory of Chemicals and Chemical Substances (PICCS)	
Europe	EEC International Cosmetics Ingredients Inventory (INCI)	absorbant/ buffering/ opacifying / additives
	EU REACH pre-registered	
	EU Inventory of Existing Commercial Chemical Substances (EINECS)	215-171-9
	German Water Hazard Class Substance List	5208
		Classification: VwVwS
	Switzerland Giftliste 1 (List of Toxic Substances)	G-2368
Canada	Canadian Domesticated Substances List (DSL)	
	WHMIS Ingredient List	
United States	ACGIH Thrshold Limit Values (TLV)	
	EPA Pesticide Inert Ingredients	
	FDA Priority-based Assessment of Food Additives (PAFA)	
	FDA Regulations	Use as colorant.
	High Production Volume Chemicals (HPV)	
	National Toxicology Program Technical Reports List	
	NIOSH Hazard, Toxicology, and Use Information	

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Magnesium oxide	e (1309-48-4)	
	NIOSH Health Hazards	
	NIOSH Recommended Exposure Limits	10 mg/m ³
	OSHA Permissible Exposure Limits	8 hour TWA: total particulates 15 mg/ m³
	Toxic Substances Control Act (TSCA) Inventory	
	Toxic Inventory Update Rule	
	TSCA Section 8A-Preliminary Assessment Information Rule (PAIR)	
Other	Health Hazards	RTECS: OM3850000
	High Production Volume Chemicals: ICCA	
	High Production Volume Chemicals: OECD	

Magnesium hyd	roxide (1309-42-8)	
Jurisdiction	List	Comment
Asia Pacific	Asia - PAC	
Australia	Australian Inventory of Chemical Substances (AICS)	
China	Inventory of Existing Chemical Substances (IECSC)	
Japan	Existing and New Chemical Substances (ENCS)	# 1-386; inorganic compounds
Korea	KECI (Chemical Inventory of Korea)	KE-22716
New Zealand	Inventory of Chemicals (NZIoC)	HSNO approval
Phillipines	Inventory of Chemicals and Chemical Substances (PICCS)	
Europe	EEC International Cosmetics Ingredients Inventory (INCI)	absorbant/ buffering
	EU REACH pre-registered	
	EU REACH registered	01-2119488756-18-0001
	EU Inventory of Existing Commercial Chemical Substances (EINECS)	215-170-3
	German Water Hazard Class Substance List	Classification: VwVwS
	Switzerland Giftliste 1 (List of Toxic Substances)	G-8166 Toxic Category 4
Canada	Canadian Domesticated Substances List (DSL)	
North America	DOT Coast Guard Bulk Hazardous Materials	
	EPA Pesticide Inert Ingredients (PII)	
	FDA Food Substances Generally Recognized as Safe (GRAS)	
	FDA Priority-based Assessment of Food Additives (PAFA)	
	High Production Volume Chemicals (HPV)	
	OSHA Permissible Exposure Limits	8 hour TWA: total particulates 15 mg/ m ³
	Toxic Substances Control Act (TSCA) Inventory	
	Toxic Inventory Update Rule (IUR)	
	TSCA Section 8A-Preliminary Assessment Information Rule (PAIR)	
	High Production Volume Chemicals: ICCA	
	High Production Volume Chemicals: OECD	

15.3. US State regulations

Magnesium Oxide (1309-48-4)	
State or local regulations	U.S. – Illinois Right-to-Know Toxic Substances List U.S. – Massachusetts Right-to-Know U.S. – Minnesota Right-to-Know U.S New Jersey Right-to-Know U.S. – Pennsylvania Right-to-Know U.S. – Rhode Island Right-to-Know
	ore. Tareas related right to ration

State or local regulations Not listed	

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MARNIN(:

This product can expose you to Lead and Nickel compounds, which are known to the State of California to cause cancer. For more information go to www.P65Warnings.ca.gov.

SECTION 16: Other information

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: 03/02/2021 Revision date Data sources : ACGIH 2019.

Chemical Inspection & Regulation Service; accessed at: http://www.cirs-

reach.com/Inventory/Global Chemical Inventories.html.

Ind. Exposure & Control Techn. for OSHA Regulated Substances - MgO (fume), March, 1989, pp. 1181-1184. Krister Forsberg and S.Z. Mansdorf, "Quick Selection Guide to Chemical Protective Clothing", Fifth Edition.

NIOSH Occupational Health Guide for chemical Substances - Vol. II, September, 1978.

REGULATION (EC) No 1272/2008 OF THE EUROPEAN PARLIAMENT AND OF THE COUNCIL of 16 December 2008 on classification, labelling and packaging of substances and mixtures, amending and repealing Directives 67/548/EEC

and 1999/45/EC, and amending Regulation (EC) No 1907/2006.

RTECS, June 1998.

Sax - 8th Ed. TSCA Chemical Substance Inventory. Accessed at

http://www.epa.gov/oppt/existingchemicals/pubs/tscainventory/howto.html.

US National Library of Medicine National Institutes of Health Haz-Map. Accessed at http://hazmap.nlm.nih.gov

Other information : None.

Abbreviations and acronyms:

ACGIH (American Conference of Government Industrial Hygienists)
ATE: Acute Toxicity Estimate
CAS (Chemical Abstracts Service) number
EC50: Environmental Concentration associated with a response by 50% of the test population.
GHS: Globally Harmonized System (of Classification and Labeling of Chemicals
LD50: Lethal Dose for 50% of the test population
OSHA: Occupational Safety & Health Administration
TSCA: Toxic Substances Control Act
TWA: Time Weighted Average

NFPA health hazard

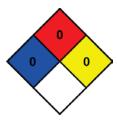
: 0 - Materials that, under emergency conditions, would offer no hazard beyond that of ordinary combustible materials.

NFPA fire hazard

: 0 - Materials that will not burn under typical fire conditions, including intrinsically noncombustible materials such as concrete, stone, and sand.

NFPA reactivity

: 0 - Material that in themselves are normally stable, even under fire conditions.



Indication of changes:

Section	Changed item	Change	Comments
15	California Proposition 65 Disclosure	Added	

SDS Prepared by: The Redstone Group

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This information is based on our current knowledge and is intended to describe the product for the purposes of health, safety and environmental requirements only. It should not therefore be construed as guaranteeing any specific property of the product.

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